## Assignment #1

Law!

- 1. What does the 1st Law of Thermodynamics entail about the relation between  $Q_{in}$ ,  $Q_{out}$ , and W for a heat engine?
- 2. Show that, if Clausius' version of the 2nd Law if false, then so is Thomson's (i.e., Kelvin's).

  Hint 1: Draw a heat engine that violates Clausius's version of the 2nd Law, and hook it up to an appropriately constructed, allowable heat engine in such a way that the product is a heat engine that violates Thomson's version of the 2nd Law.

  Hint 2: An "allowable" heat engine is a heat engine that obeys both the 1st Law and the 2nd
- 3. Show that, if Thomson's version of the 2nd Law if false, then so is Clausius'.

  <u>Hint 1</u>: Draw a heat engine that violates Thomson's version of the 2nd Law, and hook it up to an appropriately constructed, *allowable* heat engine in such a way that the product is a heat engine that violates Clausius' version of the 2nd Law.
  - <u>Hint 2</u>: An "allowable" heat engine is a heat engine that obeys both the 1st Law and the 2nd Law!